



A case study: Thermophysics characterization of tropical soils extracted in the horizons HA and HB

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Abstract: *Thermal Wave Interferometry (TWI) technique was used to access the thermal diffusivity and thermal conductivity of soils with sandy and clayey textures from Paraná land, specifically from horizons HA (5 to 25 cm) and HB (25 to 76 cm). Experimental setup involved focusing white light on soil samples in titanium wells (100-1200 μm thick), modulated at frequencies from 0.6 to 45 Hz. Results showed that sandy soils had higher thermal diffusivity and conductivity than clayey soils. Open Photoacoustic Cell (OPC) also measured the thermal diffusivity, while TWI determined thermal conductivity ranged from 0.3 to 1 $\text{Wm}^{-1}\text{K}^{-1}$, and thermal diffusivity from 1 to 9 $\times 10^{-7} \text{m}^2/\text{s}$. Specific heat was measured using Non-Adiabatic Thermal Relaxation Calorimetry (NATRAC), revealing values between 820 to 920 $\text{J/kg}\cdot\text{K}$ for HA soils and 940 to 1140 $\text{J/kg}\cdot\text{K}$ for HB soils. Gravimetric measurements provided soil particle and bulk density data.*

Keywords: Horizons. Lands. Thermal Properties. Thermal Waves.

1 Introduction

Extensive use of soils requires a comprehensive understanding of their physical, chemical, and biological properties [1, 2]. In the civil construction sector, more resistant soils are more adequate to be used as raw material. The sensing of thermal comfort comes from the thermal effusivity (e_s) contact of the materials with the environment. Gas and oil exploration often comes from the examination of soils by specialized professionals who use high-tech equipment to analyze samples

collected from the seabed [3]. The soil itself represents a complex and dynamic system that can overcome many modifications under environmental condition [4, 5]. Therefore, it very important to our knowledge the studies related to soils quality for growing crops, water retain, and mainly properties that can influence soils under pollution. Thus, environmental interventions always come to improve the sustainability and wellness of the earth resources, good and healthy crops depend on it actions [6]. Modified concrete with doping material [7] can be used seeking mechanical properties of soil when applied to concrete molding, prop wall or asphalt pavement, and some engineering are interested in better earth leveling works and draining water, for example.

Nowadays, remote sensing technical instructions are being studied by researchers and scientists of the soils because the spectral response can be found with agility and without causing damage to the environment. Remote Sensing is based on fact that soils present different absorption and reflection with the wavelength according to their physical chemical attributes [8]. Thermal regime of soils is governed by the heating of the surface from solar radiation and the transfer of heat into the interior of the soil through conduction. During the day, the surface warms up, creating a heat flow into the interior. At night, the surface cools by emitting terrestrial radiation (long-wave radiation), causing the reversal of the heat flow direction, moving from the Earth's interior to the surface. Changes in soil temperature throughout the day (temporal variations) and over depth (spatial variations) are studied using horizontal profiles [9]. Heat flux in the soil is basically dependent on its thermal conductivity (k_0), specific heat (c_{ps}), thermal effusivity ($e_s = k_s/\sqrt{\alpha_s}$), and the thermal diffusivity (α_s). Liu *et al.* studied the influence of temperature variation in nitrification of heterotrophic and autotrophic soils [10]. Carbon content, stability of aggregates and main characteristics of humic substances [11] and microbial activity [12] are more affected by varying the temperature in each soil.

Soil classes are taxonomically individualized, classified by their texture, structure and organic matter content. Sandy soils tend to have higher daily temperature amplitudes in the superficial layers and smaller in depth. This is due to the sandy soils have a higher porosity with less contact between soil particles, thus hindering the driving process. In turn, clay soils have more efficiency in conducting heat, having a lower daily thermal amplitude. Thermophysical properties of soil, such as thermal conductivity, thermal diffusivity and thermal effusivity, play an important role in the sustainability of ecosystems and the management of natural resources, because mapping thermal behavior helps in the development of agricultural practices, conservation and soil efficiency [13, 14]. For example, choosing suitable crops for different soil types and implementing conservation techniques such as ground cover and efficient supervision can be adjusted based on the thermal properties of the soil. Thermophysical properties of soil influence the response of ecosystems to climate change. Soils that better regulate temperature and humidity can help mitigate the effects of extreme climate variations, promoting the resilience of natural and agricultural systems [15].

Wang and Zhang [16] observed that soils with higher thermal effusivity generally have good water retention capacity and a denser texture, also, can help moderate temperature variations, promoting a more stable environment for plant growth and microbial activity. Thermal effusivity is particularly important for conserving soil moisture and reducing thermal stress in plants during periods of intense heat.

The main objective of this work is characterized by following thermal properties: thermal diffusivity (α_s), thermal conductivity (k_s) and thermal effusivity (e_s) of the soils with sandy textured (orders Red Argisol and Quartzarenic Neosol) and clayey (Red Latosol and Red Nitosol) from State Parana in Brazil, collected in the surface horizons HA (from 5 to 25 cm) and subsurface HB (from 25 to 76 cm) by means of the Thermal Wave Interferometry (TWI). Thermal diffusivity of Red Argisol samples were also measured by acoustic waves detection using the Open Photoacoustic Cell (OPC). Beyond photothermal and photoacoustic experiments, specific heat (c_{ps}) was obtained by the technique of Non-Adiabatic Thermal Calorimetric (NATRAC), and bulk (ρ_{bulk}) and particle (ρ_s) densities were measured according to the NBR 6508/1984 determined by Brazilian Standard. Results showed how thermophysical properties are dependent on texture and depth in tropical soils.

2 Methodology

2.1 Soils recognition

The study area was concentrated in the State of Paraná, 24°36' 0" South latitude, 51° 23' 0" west longitude, located in Brazilian territory. The state presents a humid tropical climate, temperate climate and wide vegetation, with fifty percent of its territory located at an altitude above 600m and only 3 % at an altitude below 300m. The geology of the state covers a region of transitional zone of basaltic and igneous formation.

This association of geological aspects with characteristics variables and different constituents originating from the intemperance process, being the Red Latosol and Red Argisol as predominating classes, followed by Red Nitossolo, Haplic Cambisol and Quartzarenic Neosol [17]. Figure1 presents the location of soils extraction in Paraná State of Brazil.

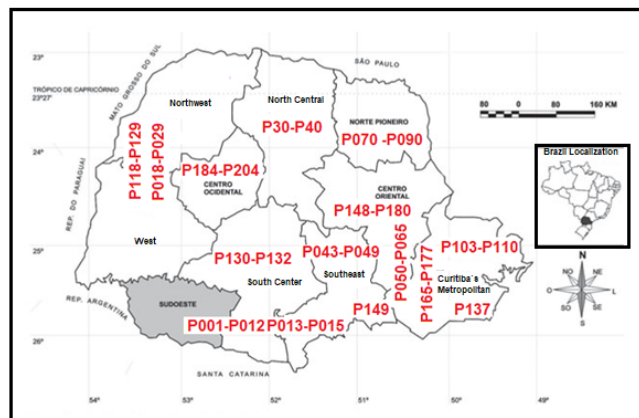


Figure 1: Regions of Parana State where soils were extracted for thermophysical analysis using photothermal methods.

For soil samples storage, collection and georeferencing of the studied sites were utilized specific materials and equipment, among which emphasize: Treaty Dutch type for collecting soil samples at the two studied depths of HA horizons (depths between 5 and 25 cm), HB horizons (depths between 25 and 76 cm), plastic bags and other materials used for packaging, notebooks to register and receive GPS brand Trimble (model GeoExplorer 3) with differential correction for georeferencing points in the field, photographic camera to visual documentation of work area images. Previously the collect, Cezar and coworkers [18] conducted previous studies supported by geographic information system (SIG) with a logistical purpose for optimizing sampling. Sampling density was one point in each 250.000m². The collected samples were sent to the Fertility Laboratory of Agronomy Department of the Universidade Estadual de Maringá where they were air-dried and sifted in mesh of 2mm to obtain Dry Earth in the Air (TFSA).

Following the routine physical-chemical analysis described by Cezar et al. [18], the samples were used to identify soil classes. All the material was first dried and then subjected to manual shredding. Chemical treatment was done through the addition of 12.5 % sodium hypochlorite for taking off organic matter and NaOH was used for extraction of sand, silt and clay levels. The sand fraction was segregated by sieving at 53µm of diameter, and the clay fraction was segregated after sedimentation of the silt fraction according to the Stoke's law. Next separation the clay fraction was flocculated by MgCl₂ and overage mineral salts were removed by successive washes. In following the clay was dried in an oven at 50 °C for 48 h and next the clay granules were hermetically homogeneous in the mortar. The hardest samples were processed using a Mimoso-brand machine for medium disruption and homogenization. The pulverized raw material was initially sun-dried and then placed in a greenhouse at 60°C for 24 hours to remove excess moisture. Fine blending was done in the mortar with help of the pistil (Soils Laboratory LECIV/CCT/UENF). The remaining powder in each layer, was collected in sufficient amount to carry out the tests (granulometric analysis, liquidity limit, plasticity limit). Levels organic carbon and Fe₂O₃ were obtained according to Brazilian Agricultural Research Company (EMBRAPA) standards.

Table 1 shows granulometry analysis of the samples. Among the studied samples, forty-one samples are Red Latosols (RL) and Yellowy Red Latosols (YL), twenty-one belong to the class of the Red Argisol (RA), twelve of the Haplic Cambisols (HC), ten extracted samples are Quartzarenic Neosols (QN) and two samples are of the class Red Nitosol (RN). Four samples studied classes in this work represent 70 % of the pedological territory of the state of Paraná / Brazil. Its clayey nature, silty and sandy depend on the predominance of the diameter of its grains. The clayey soils are those that present in their composition the predominance of clay above 35 %, and very clayey when a concentration of clayey grains is higher than 60%. To be considered as sandy soil, the sand composition must be higher than 70%, to be conceptualized as medium sandy soil, the soil must be formed between 15% and 70% of sand grains, and finally, silty soils are those that have at least 70% of silts grains.

| Samples Identification | Classification | | | % | | | Predominant Texture |
|------------------------|----------------|----------|------------------|------------|-----------|-----------|---------------------|
| | Code | Suborder | Horizontal Layer | Thick Sand | Thin Sand | Silt Clay | |
| 391P141 | RA | HA | 60 | 04 | 07 | 29 | very sandy |

| | | | | | | | |
|----------|----|----|----|----|----|----|------------|
| 439P054 | RA | HA | 56 | 11 | 04 | 29 | very sandy |
| 450P097 | RA | HA | 52 | 03 | 42 | 03 | very sandy |
| 458P061 | RA | HA | 66 | 02 | 21 | 11 | very sandy |
| 466P198 | RA | HA | 66 | 09 | 06 | 19 | sandy |
| 620P074 | RA | HA | 46 | 25 | 02 | 27 | sandy |
| 633P072 | RA | HA | 74 | 13 | 02 | 11 | Sandy |
| 523P086 | RA | HA | 30 | 48 | 02 | 20 | Sandy |
| 512P152 | RA | HA | 36 | 27 | 03 | 34 | very sandy |
| 616P084 | RA | HA | 32 | 16 | 27 | 25 | very sandy |
| 511P190 | RA | HA | 31 | 30 | 06 | 33 | very sandy |
| 613P076 | RA | HA | 21 | 31 | 13 | 35 | very sandy |
| 402P097 | RA | HB | 52 | 03 | 03 | 42 | very sandy |
| 422P141 | RA | HB | 51 | 04 | 04 | 41 | very sandy |
| 436P095 | RA | HB | 57 | 04 | 03 | 36 | very sandy |
| 441P059 | RA | HB | 65 | 17 | 02 | 16 | Sandy |
| 624P071 | RA | HB | 51 | 18 | 04 | 27 | very sandy |
| 496P198 | RA | HB | 44 | 13 | 10 | 33 | very sandy |
| 507P211 | RA | HB | 31 | 30 | 04 | 35 | very sandy |
| 424P056 | HC | HA | 48 | 16 | 05 | 31 | very sandy |
| 430P148 | HC | HA | 19 | 27 | 05 | 49 | Clay |
| 443P162 | HC | HA | 27 | 06 | 10 | 57 | Clay |
| 445P055 | HC | HA | 21 | 35 | 05 | 39 | Clay |
| 455P099 | HC | HA | 49 | 07 | 04 | 40 | Clay |
| 615P164 | HC | HÁ | 19 | 17 | 03 | 61 | very clay |
| 429P159 | HC | HB | 41 | 06 | 15 | 38 | Clay |
| 437P099 | HC | HB | 49 | 06 | 04 | 41 | Clay |
| 448P055 | HC | HB | 15 | 40 | 03 | 42 | Clay |
| 476P166 | HC | HB | 27 | 22 | 16 | 35 | Clay |
| 510P152 | HC | HB | 19 | 30 | 04 | 47 | Clay |
| 514P065 | HC | HB | 11 | 10 | 13 | 66 | very clay |
| 401P138 | QN | HA | 32 | 31 | 02 | 35 | Clay |
| 621P075 | QN | HA | 37 | 23 | 02 | 38 | Clay |
| 622P078 | QN | HA | 24 | 20 | 03 | 53 | Clay |
| 442P188 | QN | HA | 40 | 18 | 05 | 37 | Clay |
| 444P151 | QN | HA | 37 | 06 | 10 | 47 | Clay |
| 402P135 | QN | HA | 35 | 17 | 08 | 40 | Clay |
| 421P187 | YL | HA | 15 | 10 | 10 | 65 | very clay |
| 504P079 | YL | HA | 11 | 16 | 03 | 70 | very clay |
| 508P089 | YL | HA | 23 | 08 | 03 | 66 | very clay |
| 449P180 | QN | HA | 45 | 11 | 05 | 39 | Clay |
| 491P183 | YL | HA | 11 | 11 | 09 | 69 | very clay |
| 610P068 | YL | HA | 21 | 13 | 02 | 64 | very clay |
| 483P166 | YL | HÁ | 21 | 13 | 02 | 64 | very clay |
| 604P82 | QN | HA | 22 | 16 | 24 | 38 | Clay |
| 628P080 | QN | HA | 32 | 17 | 16 | 35 | Clay |
| 444P151 | QN | HA | 37 | 06 | 10 | 47 | Clay |
| 510P152 | QN | HA | 34 | 27 | 03 | 36 | Clay |
| 417P060 | YL | HA | 12 | 18 | 02 | 68 | very clay |
| TAMARANA | QN | HA | | | | | |
| TAMARANA | QN | HA | | | | | |
| TAMARANA | QN | HA | | | | | |
| TAMARANA | QN | HA | | | | | |
| 396P138 | QN | HB | 50 | 06 | 04 | 40 | Clay |
| 397P136 | QN | HB | 25 | 06 | 18 | 51 | Clay |
| 412P051 | YL | HB | 15 | 07 | 02 | 76 | very clay |
| 418P189 | QN | HB | 44 | 16 | 04 | 36 | Clay |

| | | | | | | | |
|---------|----|----|----|----|----|----|------------|
| 451P096 | QN | HB | 46 | 06 | 03 | 45 | Clay |
| 489P068 | QN | HB | 42 | 13 | 04 | 41 | Clay |
| 495P210 | QN | HB | 44 | 15 | 06 | 35 | Clay |
| 513P090 | YL | HB | 13 | 04 | 08 | 75 | very clay |
| 525P88 | YL | HB | 13 | 15 | 02 | 70 | very clay |
| 612P33 | YL | HB | 11 | 07 | 04 | 78 | very clay |
| 393P188 | QN | HB | 40 | 12 | 05 | 43 | Clay |
| 394P185 | QN | HB | 26 | 09 | 30 | 35 | Clay |
| 406P180 | QN | HB | 31 | 22 | 03 | 44 | Clay |
| 490P179 | QN | HB | 13 | 22 | 09 | 56 | Clay |
| 399P135 | YL | HB | 24 | 11 | 04 | 61 | very clay |
| 408P136 | QN | HB | 25 | 06 | 18 | 51 | Clay |
| 519P089 | QN | HB | 26 | 27 | 04 | 43 | Clay |
| 518P183 | YL | HB | 15 | 18 | 02 | 65 | very clay |
| 619P180 | YL | HB | 20 | 07 | 12 | 61 | very clay |
| 486P035 | RN | HA | 11 | 08 | 11 | 70 | very clay |
| 486P035 | RN | HB | 11 | 08 | 11 | 70 | very clay |
| 503P066 | RQ | HA | 72 | 12 | 02 | 14 | Sandy |
| 309P139 | RQ | HÁ | 64 | 17 | 04 | 15 | Sandy |
| 413P094 | RQ | HA | 73 | 11 | 01 | 15 | Sandy |
| 600P070 | RQ | HA | 49 | 06 | 12 | 33 | very sandy |
| 411P139 | RQ | HB | 65 | 13 | 03 | 19 | Sandy |
| 475P060 | RQ | HB | 70 | 08 | 03 | 19 | Sandy |
| 454P094 | RQ | HB | 64 | 15 | 02 | 19 | Sandy |
| 414P093 | RQ | HB | 69 | 07 | 05 | 19 | Sandy |
| 607P070 | RQ | HB | 51 | 01 | 03 | 45 | very sandy |
| 494095 | RQ | HB | 09 | 08 | 03 | 80 | very clay |

*Mineral soil, composed mainly of organic materials.

Table 1: Particle size analysis of Paraná soil samples.

After that, the soil samples were macerated in the agate mortar for that a large part of the larger agglomerates and grains were fragmented and sieved in 45 μm about 20 minutes. This work was needed due to mathematical photothermal models require better flatness and homogeneity of the material, in addition to reducing the effects of humidity and roughness. Since soil samples were porous, with irregular granules of different sizes, such special care was required. Soil samples were stored in closed acrylic pots to minimize the exchange humidity with the environment. Hematite ($\alpha\text{-Fe}_2\text{O}_3$) and sandy samples were also sieved and macerated to prepare them for calorimetric and gravimetric measurements.

2.2 Mapping soil thermophysical properties by thermal wave technique

Thermal Waves Interferometry (TWI) is a remote and nondestructive technique used to thermophysical characterization of solids and gases and to detect defects in layers and coatings. Bennet and Patty's model [19] foresees that while solid surface is excited with modulated and focused light, thermal waves are generated in the material. These waves propagate by means of thermal conduction to substrate-sample and sample-gas interfaces, where they are reflected successively until they are completely attenuated by e^{-x/μ_s} , in which $\mu_s = \sqrt{\frac{2\alpha_s}{\omega}}$ is known as the sample's thermal diffusion length, α_s , is the thermal diffusivity, and ω is the light modulation angular frequency. The main effect of the reflection of thermal waves in the sample is the fluctuation of the surface temperature, θ . The surface temperature oscillation is a complex quantity mathematically depicted by the integration of the sum of the reflection's contributions at the interfaces in the whole sample thickness, l_s . Equation (1) shows that temperature fluctuation behaves faultlessly as an evanescent space wave, decreasing exponentially to points distant from heat sources.

$$\theta = \frac{I_o \beta T_g}{4k_s \sigma_s (1 - R_b R_g e^{-2\sigma_s l_s})} \left[\frac{1 - e^{-(\beta + \sigma_s) l_s}}{\beta + \sigma_s} + \frac{R_b e^{-2\sigma_s l_s}}{\beta - \sigma_s} (1 - e^{-(\beta - \sigma_s) l_s}) \right]. \quad (\text{Equation 1})$$

In which, I_o , is the light beam initial intensity, β is the opacity coefficient of sample, $\sigma_s = \frac{(1+i)}{\mu_s}$, is the complex sample thermal diffusion coefficient, $R_b = \frac{e_s - e_b}{e_s + e_b}$, $R_{bg} = \frac{e_s - e_g}{e_s + e_g} \sim 1$, are respectively the reflection coefficients at the substrate-sample and sample-gas interfaces, associated to the thermal effusivity of the sample, e_s , substrate, e_b , and gas, e_g , respectively, and, $T_g = \frac{e_s}{e_s + e_g}$, is defined as the transmission coefficient of the sample for the gas [20]. Considering that the opaque sample is normalized by thermal thickness, $\frac{l_s}{\mu_s}$, phase shift, φ , between a reference and samples and is set by:

$$\varphi = \tan^{-1} \left\{ \frac{R_b (1 + R_g) e^{-2\frac{l_s}{\mu_s}} \sin(2\frac{l_s}{\mu_s})}{1 - R_g \left(R_b e^{-2\frac{l_s}{\mu_s}} \right)^2 + R_b (1 - R_g) \cos(2\frac{l_s}{\mu_s})} \right\}. \quad (\text{Equation 2})$$

Phase shift is independent of the magnitude of the absorbed and reflecting energy in the surface. Phase shifts are influenced by the material thickness due to the super damped and harmonic nature of thermal waves. In practice, measurements are taken over a wide frequency range, which

influences the depth of heat penetration of the thermal wave in the sample until reaching $\varphi \cong 0$, finding the thermally thick regime of the wave. The exchange between the thermally thin and thick regimes thermal wave are observed at the points of inflection of the curves, that is at, $\frac{l_s}{\mu_s} = 0.75$ for signal amplitude, and $\frac{l_s}{\mu_s} = 1.5$ to phase [19]. Simulation of Eq. (2) is shown in Figure 2 using several values of R_b from $R_b=-1$ (shift $\sim +45$) to $R_b=+1$ (shift ~ -45).

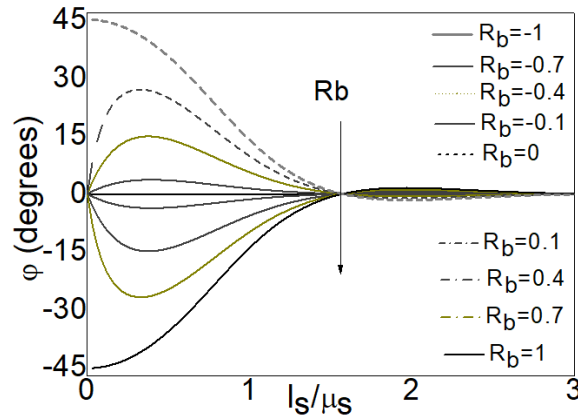


Figure 2: Normalized phase shift thermal wave of the opaque sample in range $-1 \leq R_b \leq 1$.

TWI setup is presented in Figure 3. The system consists of a Xenon arc lamp emitting white light (Oriol Instruments, model 68820), a chopper (Stanford Research Systems, model 540) modulating the light, a thermopile (Oriol Instruments, model 71755) with a Germanium (Ge) window with transmission in 90 % at range (8-14) μm to read the sample average surface temperature fluctuation, a lock-in amplifier (Stanford Research Systems, model 850) which uncouple phase and amplitude signal monitored with the thermopile at the reference frequency of the chopper, and a X-Y stage driven by a bipolar stepper motor (API GETTYS, model 230-6102BN) accurately 10 steps/mm and course of 10 cm for the X-Y stage, limited by two path final sensors (PAI 0221) operating in the clockwise and counterclockwise directions. A computer with an acquisition program was employed to receive, update and store the preamplifier data, communicate with the power supply of the step motor and control the voltage received by path final sensors. Two flat-convex lenses BK7 with focal length f of 10 and 15 cm, a flat mirror and a flat-convex lens BK7, $f = 5$ cm were used to collimate light beam and helping to focus the light on the sample. Scheme is shown in Figure 3a.

Samples already macerated and sieved were placed with the aid of a spatula in wells of varying thicknesses, between 100 to 1200 μm , onto a titanium (Ti) substrate machined with the thickness $l_b = 2$ mm. As shown in Figure 3b, the soil was deposited in the wells and smoothed by a blade to ensure flatness of the surface in the desired thickness. Substrate was put on the X-Y stage moving from 0 to 100 mm while the sample's face and of the substrate illuminated by a fixed light modulation frequency, f .

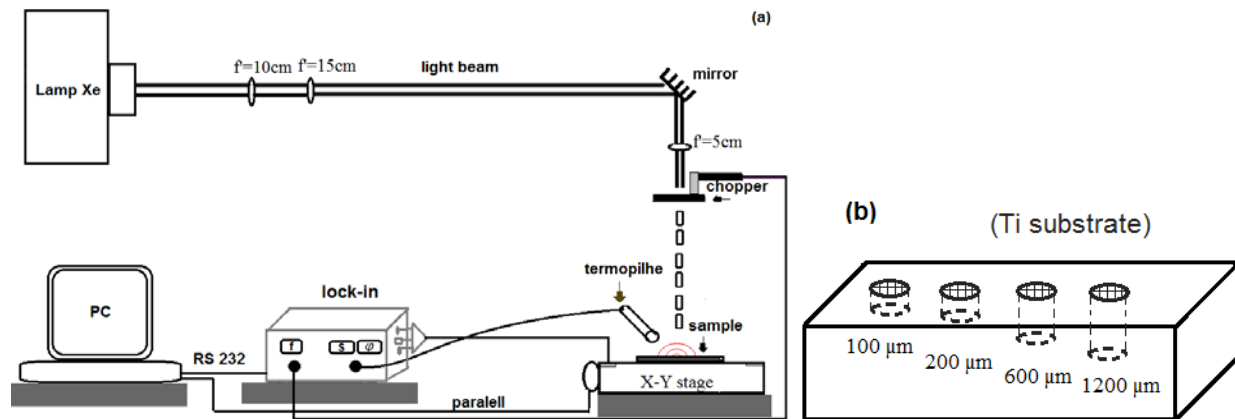


Figure 3a: TWI experimental setup and **Figure 3b:** Soil samples put titanium substrate.

During the experiment, the thermopile monitored the surface temperature fluctuations of both the substrate and the sample at regular intervals of $0.5\ \text{mm}$ (five steps). The procedure involved taking an average of 10 readings per point, with measurements performed at twice the time constant of the lock-in amplifier, $t=3\text{s}$, updating the position during acquisition. These data were recorded in tables per position, signal and phase, and then plotted in graphs (signal vs. position and phase vs. position). Measurements were repeated within the frequency range of 0.6 to $45\ \text{Hz}$. The wells in which each sample was deposited had a diameter of $6\ \text{mm}$ and were spaced $10\ \text{mm}$ apart on the substrate. During laboratory tests, the ambient temperature was adjusted to 25°C , and the relative humidity of the air was monitored using a thermo-hygrometer (ICEL Manaus, model HT-4000). The samples were measured again after being dried in a greenhouse for 24 hours (FANEM LTDA, model 311 CG) at approximately 105°C . Thermophysical properties of the soils measured immediately were: thermal diffusivity (α_s), thermal effusivity (e_s), and indirectly thermal conductivity (k_s).

2.3 Thermal diffusivity investigation by photoacoustic method

2.3.1 Samples preparation

To prevent soil compaction and ensure uniform sample dimensions, sample holders were made from $12\ \mu\text{m}$ thick aluminum foil, used as layers for thermal coupling due to its ease of acquisition and handling. In this setup, the coefficient of thermal diffusion is greater than the thickness of the metal sheets ($\mu_0 > l_0$), given that the cutting frequency of aluminum is around $7\ \text{kHz}$. Consequently, for measurements up to $220\ \text{Hz}$, the aluminum layers are considered thermally thin. Figure 4 illustrates the sample holders used for storing soils. A vacuum grease was employed to seal each sample holder. A total of twelve samples were prepared, with two samples of each class extracted from horizons HA and HB. Each sample had a diameter of $10\ \text{mm}$ and a thickness ranging from 221 to $284\ \mu\text{m}$.

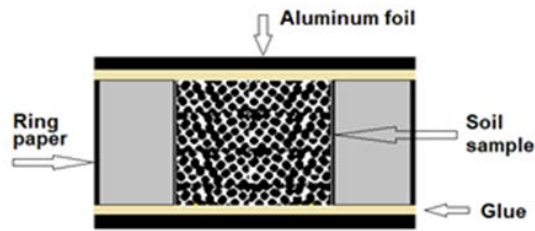


Figure 4: Soil's sample support.

2.3.2 Thermal diffusivity

Experimental setup used for measuring thermal diffusivity is shown in Figure 5.

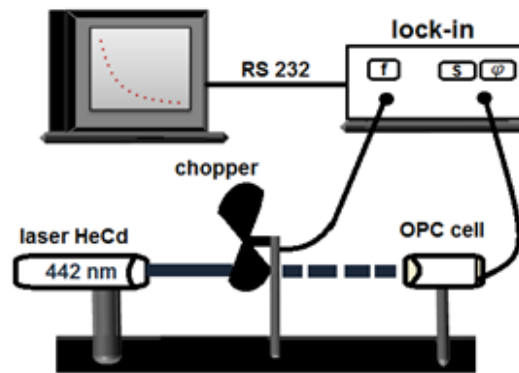


Figure 5: Experimental OPC setup for thermal diffusivity measurement of soils.

Vacuum grease was used to seal the sample in the cell. A modulated light beam was directed at the back of the sample, causing heat to diffuse through the material until it reached the other side. A He-Cd laser (K Series, KR KiMMON, model IK5652R-G) with a power of 120 mW at 442 nm was used as the excitation source, modulated at frequencies ranging from 10 to 220 Hz. As the samples heated due to the absorption of modulated light, pressure oscillations within the cell, matching the frequency of the chopper, were detected by an electret microphone (model WM-61A, Panasonic) in the cell's OPC. The signals were recorded by a lock-in amplifier in terms of phase and amplitude, with values stored on a microcomputer. According to the model proposed by Rosencwaig and Gersho [21], considering thermal diffusion as the primary mechanism for generating the photoacoustic signal, the equation governing the pressure fluctuations in the air chamber is described by:

$$\partial P = \frac{\gamma P I_0}{2\pi l_g k_s f \theta_{amb} \sinh(l_s \sigma_s)} \sqrt{\alpha_s \alpha_g} e^{i(\omega t - \frac{\pi}{2})}. \quad (\text{Equation 3})$$

In Eq. (3), P is the ambient pressure, α_s is the surrounding gas thermal diffusivity and l_s is the gas thickness. Photoacoustic signal corresponds to the non-temporal part of the pressure fluctuation ∂P . Thermal diffusivity can be calculated according to the thermal regime that sample

is found. In thermally thin regime ($\sinh(x) \sim x$), pressure fluctuation is proportional to $f^{3/2}$, like is represented by Equation. (4)

$$\partial P = \frac{\gamma P I_o}{(2\pi)^{3/2} l_g l_s k_s f \theta_{amb} f^{3/2}} \sqrt{\alpha_g \alpha_s} e^{i(\omega t - \frac{3\pi}{4})}. \quad (\text{Equation 4})$$

On the other hand, in frequencies above the critical frequency, (f_c), the sample is termed as thermally thick and the equation follows the approximation $\sinh(x) \sim e^{-x}$, then, $\partial P \approx \frac{e^{-b\sqrt{f}}}{f}$:

$$\partial P = \frac{\gamma P I_o}{(2\pi)^{3/2} l_g l_s k_s f \theta_{amb} f^{3/2}} \sqrt{\alpha_g \alpha_s} e^{i(\omega t - \frac{\pi}{2} - \frac{l_s}{\mu_s})} \frac{e^{-b\sqrt{f}}}{f}. \quad (\text{Equation 5})$$

In which, the constant b is given by $b = l_s \sqrt{\frac{\pi}{\alpha_s}}$. Photoacoustic signal amplitude for a sample terminally thick decrease exponentially [22] with the modulation frequency, i.e., $S_{PA} \propto \frac{e^{-b\sqrt{f}}}{f}$. In OPC, α_s is obtained from the adjustment of experimental results from coefficient (b) in exponential argument ($-b\sqrt{f}$). The cut frequency found for the soils was around 7 Hz, i.e., as all the measurements in this work are comprised from 10 to 220 Hz, the thermally thick regime was guaranteed.

2.4 Specific heat measurement: thermal relaxation calorimetry at ambient temperature

Thermal relaxation calorimetry is a technique that measures temperature fluctuations as a function of time following the application of a heat pulse [23]. A detailed description of this theory is presented by Barucci *et al.* [24].

In the tests, the initially empty substrate was heated by a laser beam during 60s until temperature of 25°C. Soon after, the beam light was interrupted by a shutter and the heat was dissipated by a silver substrate during the same time interval of the heating cycle. Temperature variation between the thermal reservoir and the substrate was read by a thermocouple K-type coupled to the substrate and connected to a nano-voltmeter (Keithley, model 2182) which measured the potential difference related to the temperature difference as a function of time transmitting the data to a computer. Temperature monitoring of thermal reservoir was performed by a temperature controller (LakeShore, model 340).

Forthwith, an aluminum substrate of known mass (24.5 mg) was fixed on the silver substrate and the measurements of heating and thermal relaxation cycles were retrying, so that power absorbed by the system was 20mW. Tests shown that the fused silica specific heat experimentally found was equivalent to $(734 \pm 5) \text{ J.Kg}^{-1}.\text{K}^{-1}$, exhibiting a relative percentage error lower than 0.2% to the literature. After calibration, approximately 12 mg of the soil samples, shown in Table 2, and sand and hematite samples were measured by NATRAC. All measurements masses were made on digital scale balance (AND, model GH202-50 μg precision).

2.5 Bulk density and particle density

Soil density can be understood in two ways: bulk ($\rho_{s\text{bulk}}$) and particle ($\rho_{s\text{part}}$). Bulk density is defined as the ratio of the dry soil mass (m_{ss}) in the greenhouse after 24 hours drying at least, 105 °C, per volume (V_t) filled by the soil in a recipient including solid and porous fraction:

$$\rho_{s\text{ bulk}} = \frac{m_{ss}}{V_t}. \text{ (Equation 6)}$$

Particle density is equivalent to the density of the soil based on the volume of the solid fraction. The particle density was determined by measuring the amount of ethyl alcohol required to fill the pipette to 25 ml, and the results were calculated accordingly.

$$\rho_{s\text{ part}} = \frac{m_{ss}}{V_l} \rho_l. \text{ (Equation 7)}$$

In which, $\rho_l = 789 \text{ Kg. m}^{-3}$ is the ethyl alcohol density.

About 10g of the soil samples and sandy were weighed and taken to the greenhouse (FANEM LTDA, model 311 CG) to dry at temperature 105°C. The samples were again weighed, and their masses recorded. Gravimetric humidity was calculated according to the NBR 6508. Following that, the samples were put in the volumetric pipette with the helpfulness of a funnel and bulk density was compute by Eq. (6). Soils specific mass was determined through addition of hydrated ethyl alcohol 92.8°ppm grade putting slowly until it reaches the mark of 25 ml in the meniscus of the pipette. The liquid volume (V_l) was transferred to another pipette and the soil particle density was calculated by Eq. (7).

3 Results and Discussion

3.1 Thermophysical properties mapping by thermal wave technique

Although photothermal techniques are well-known and applied in the characterizations of the natural porous materials, as hydroxyapatite [18, 25, 26], analytical methods for soil recognition, as carbon content and minerals, are very efficient [27-31], and effects with addition of elements in the soil already analyzed, for example, addition of selenide in soil adsorption to either increase its fertility [32], the actual work is a pioneer in the characterization and thermal mapping of Brazilian tropical soils by Thermal Waves Interferometry technique. The 85 samples were measured and characterized by TWI, as shown in Table 1. Among photothermal techniques, TWI presents some advantages to assess thermophysical properties of the soils: samples do not have to be compacted, readings are done in real time, technique has good suitability for porous and heterogeneous materials, thermal analyzes are fast and important properties can be determined [33, 34].

Initially, the signals and phases of the samples were recorded during the frequency scans. It was observed that, starting from 15 Hz, the phase difference among all thicknesses was completely damped. Next, average phase (φ) were plotted as a function of normalized thickness with frequency $l_s\sqrt{f}$. Characteristic curve of the normalized phase at the highest modulation frequency observed in the measurements is presented. For a more comparative approach, consider the phases obtained at a 3 Hz frequency for samples RL393P188HB, QN454P039HB, and RL444P151HA,

respectively, presented in Figure 6. Measured phase varies from sample to sample based on their characteristics.

The dashed line indicates a measurement on the substrate. Black lines drawn on plateaus represent the average phase found at the respective sample thicknesses measured at a fixed frequency. Soil porosity causes the phases to fluctuate on the plateau because the measurement is made on the irregular grains of the soil along a 6 mm extension of each sample thickness.

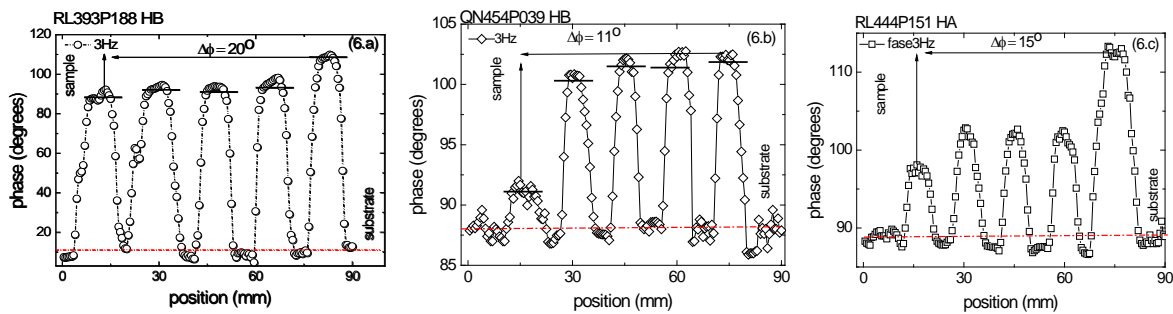


Figure 6a: Experimental phase shift data obtained for soil RL393P188HB; **Fig. 6b:** for QN454P039HB, and **Fig. 6c:** for soil RL444P151HA. All experiments used laser scanning on samples.

The determination of thermal diffusivity and reflection coefficient were obtained from the characteristic curves of the normalized wave lag by thicknesses of scans over a wide frequencies range. Fittings are made using the Eq. (2). Some characteristic curves are displayed in Figures 7a, 7b and 7c to the samples LE393P188HB, RQ454P039HB and LE444P151HA, which showed phase shifts of 19°, 11° and 11°, respectively, with the reflection coefficient, $R_b = -(0.50 \pm 0.02)$, $R_b = -(0.29 \pm 0.02)$, and $R_b = -(0.30 \pm 0.02)$.

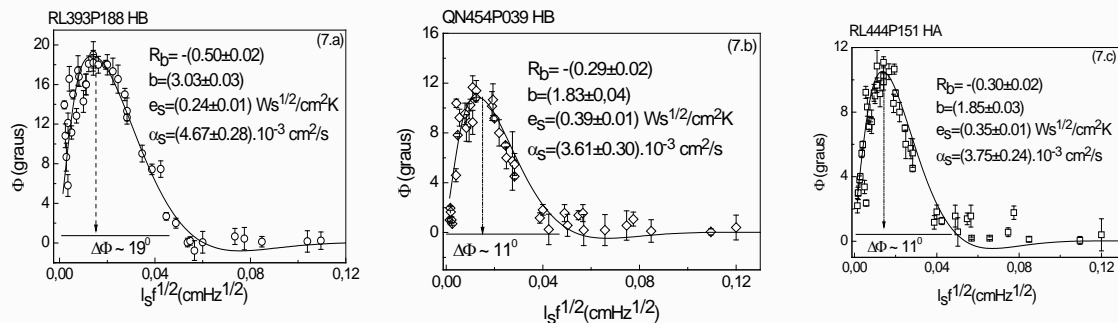


Figure 7: Fitting of the data using Eq. (2) of soils **Fig. a:** RL393P188HB; **Fig. b:** QN454P039HB, and **Fig. c:** RL444P151HA, all using the TWI phase shift data.

Negative sign of the coefficients is explained by the fact that the soils are thermally less conductive than the titanium substrate $e_b = 7172 W \cdot \sqrt{s} \cdot m^2 \cdot K^{-1}$ [20]. Thermal effusivity, e_s , is

calculated by means of the expression $e_s = e_b \left(\frac{1+R_b}{1-R_b} \right)$, and thermal conductivity, $k_s = e_s \sqrt{\alpha_s}$, was also calculated. In the curves shown in Figure 7, the characteristics of the results indicate a more pronounced phase shift and sharper width in the argisol sample extracted from the subsurface horizon. Experimental results from other samples were fitted using the phase shift model of the opaque sample. Results, which describe the average values of the thermophysical properties of studied soil classes and extraction horizons, are shown in Figures 8a, 8b, and 8c.

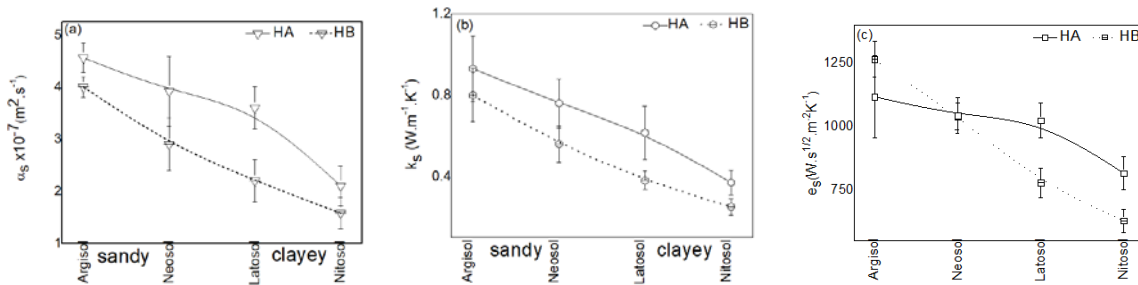


Figure 8a: Group of soils average thermal diffusivity measured by TWI method, one note that slope is decrescent from “sandy” toward “clayey” soils, **Fig. 8b:** Soils thermal conductivity and **Fig. 8c:** Soils thermal effusivity. Both stratus HA and HB follow a similar behavior for the three thermophysics properties.

Given that each type of soil has specific physicochemical attributes, it is essential to note that thermal properties can be influenced by various factors, such as texture and chemical composition. Results from TWI indicate that the mineral structure affects the thermal diffusivity and thermal conductivity of soil. In sandy soils, around 60% of the mineral composition consists of silicates, whereas in clay soils, at least 35% of the mineral content comprises iron and aluminum oxides. Silicate minerals conduct and diffuse heat more intensely and quickly than ferric oxides. Heat transfer in the surface layers of the soil, resulting from thermal disturbances due to direct solar irradiation and the incorporation of organic matter into the horizon HA, may slightly increase the thermal transport properties of the soil. This is because each studied layer of soil has a specific composition. Deep layers are strongly stratified. Figures 8a and 8b show that the thermal diffusivity was established in the range from 3 to 5 x 10⁻⁷ m².s⁻¹ in thick textured soils, and between 1.6 and 3.6 x 10⁻⁷ m².s⁻¹ for clay soils, and thermal conductivity was set in the range from 0.56 to 0.93 W.m⁻¹.K⁻¹ for the samples at the classes of the Red Argisol and Quarzarenic Neosol, and between 0.25 and 0.46 W.m⁻¹.K⁻¹ for soils Red Nitosol and Red Latosol. Thermal diffusivity controls the rate in which the thermal waves propagate through soil layers. The absence of water impedes heat flow in the soil, making it difficult for the topsoil to transfer heat to the lower layers.

Average values of thermal diffusivity are in good agreement with the range reported by other studies that used harmonic and logarithmic models to analyze French soils. Thermal conductivity of soil depends on its structure, moisture, and texture. The granule size in clayey soils is smaller than 2 μm, whereas in clay soils, the diameters can reach up to 50 μm. Smaller diameter soils require more particles to create porosity, which increases the resistance of these particles. In clayey soils, there is good particle contact and improved water storage, which inhibits heat diffusion and affects the heat transfer process.

This phenomenon suggests that sandy soils heat up faster on the surface when exposed to the same irradiation. The increase in k_s with sand content is therefore a consistent result. Results indicate that soil tends to diffuse more heat at the surface and at shallower depths.

Thermal effusivity (Figure 8c) shows the same behavioral tendency of thermal transport properties in relation to soil textures. Soil is a thermal insulation, which directly influences the parameter of heating regulation on the contact surface, that is, thermal effusivity. For the Parana soils, the thermal effusivity remained determined within the limits of 630 until 1300 $W.\sqrt{s}.m^{-2}.K^{-1}$. Due to its low thermal effusivity, the soil can retain heat for a longer period, which contributes to slower surface temperature fluctuations. This helps to mitigate the effects of sudden changes caused by climate variation, and daily cycles to which soils are constantly exposed. Soils have a hydrophilic nature due to their mineral composition and aqueous fraction. Experiments demonstrated that the gravimetric moisture, which represents the minimum soil moisture after drying processes in the greenhouse, was lower than 6% in all measurements.

Soil temperature is strongly influenced by its thermophysical parameters. To opaques samples ($\beta l_s \gg 1$) the superficial temperature equation, Eq. (1), can be simplified by being rewritten as:

$$\theta = \frac{I_o}{2k_s\sigma_s} \left[\frac{1 + R_b e^{-2l_s\sigma_s}}{1 - R_b R_g e^{-2l_s\sigma_s}} \right]. \text{ (Equation 8)}$$

By adopting a thickness of 100 μm and a frequency of 3 Hz, the electromagnetic wave intensity reaching the sample (2 W) was measured using a power sensor. Thermal conductivity and thermal diffusivity obtained from thermal wave analysis are presented in Table 2, which shows the real part of the heating for the following samples. The heating in the samples exceeded 2 Kelvins in clay soils and was less than 1.7 Kelvins in sandy soils. Experiments did not show significant temperature changes at different soil depths.

| Sample | $R_b \pm \Delta R_b$ | $\alpha_s \pm \Delta\alpha_s$ ($10^{-7} m^2.s^{-1}$) | $k_s \pm \Delta k_s$ ($10^{-1} W.m^{-1}.K^{-1}$) | $\theta \pm \Delta\theta$ (K) |
|--------------|----------------------|--|--|-------------------------------|
| RN486P035-HA | $-(0.64 \pm 0.04)$ | 1.56 ± 0.33 | 6.31 ± 0.15 | 2.04 ± 0.10 |
| YL508P089-HA | $-(0.64 \pm 0.03)$ | 1.55 ± 0.16 | 6.37 ± 0.09 | 2.01 ± 0.11 |
| RA620P074-HA | $-(0.52 \pm 0.03)$ | 2.16 ± 0.34 | 10.47 ± 0.23 | 1.44 ± 0.09 |
| RQ413P084-HA | $-(0.58 \pm 0.03)$ | 1.99 ± 0.30 | 8.61 ± 0.16 | 1.69 ± 0.10 |
| RN494P035-HB | $-(0.63 \pm 0.03)$ | 1.31 ± 0.39 | 5.84 ± 0.13 | 2.01 ± 0.11 |
| YL525P088-HB | $-(0.69 \pm 0.03)$ | 1.37 ± 0.40 | 4.93 ± 0.12 | 2.44 ± 0.13 |
| RA624P071-HB | $-(0.52 \pm 0.02)$ | 2.02 ± 0.25 | 10.19 ± 0.16 | 1.44 ± 0.10 |
| RQ414P093-HB | $-(0.55 \pm 0.03)$ | 1.68 ± 0.24 | 8.48 ± 0.14 | 1.57 ± 0.09 |

Table 2: Thermal properties found by means of surface heating.

3.2 Comparison between thermal diffusivity obtained by OPC and TWI for argisols

Twelve dry samples of Argisol extracted from the horizons HA and HB were measured using a Photoacoustic Cell. Figure 9 shows a log-log graph of the Photoacoustic signal magnitude versus

frequency for the Argisol sample 512P212 collected from horizon HB. Thermal diffusivity was determined by fitting the photoacoustic signal with Eq. (5). Table 3 indicates that the thermal diffusivity ranged from 6.5 to $9.5 \times 10^{-7} \text{ m}^2 \cdot \text{s}^{-1}$, with approximately 10% relative uncertainty, and no significant behavioral differences between horizons HA and HB. The best fit was found in the frequency interval from 57 to 120 Hz. Above 120 Hz, the linear behavior of the signal was not obtained due to the influence of thermoelastic and photobaric effects.

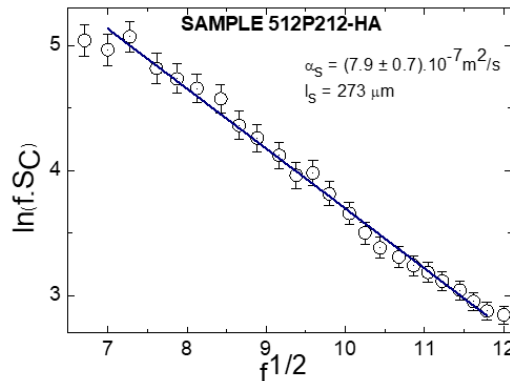


Figure 9: Linear fitting of OPC signal data to Argisol having 273 μm thickness. Sample 512P212HA. Typically, the OPC experiment allows the thermal diffusivity value of soil sample, and for the Argisol, it was found $\alpha_s = 7.9 \pm 0.7 \times 10^{-7} \text{ m}^2 \cdot \text{s}^{-1}$.

| Sample Code | Layer Horizontal | $l_{\text{sampleport}}^*$ μm | l_{glue}^* μm | l_s^* μm | α_s ($10^{-7} \times \text{m}^2 \cdot \text{s}^{-1}$) |
|-------------|------------------|--|--------------------------------------|--------------------------|--|
| 439P054 | HA | 307 | 36 | 247 | 9.4 ± 0.8 |
| 450P097 | HA | 291 | 30 | 237 | 8.7 ± 0.6 |
| 512P212 | HA | 320 | 23 | 273 | 7.9 ± 0.7 |
| 523P086 | HA | 286 | 20 | 242 | 6.7 ± 0.2 |
| 613P076 | HA | 272 | 27 | 221 | 6.3 ± 0.4 |
| 616P084 | HA | 282 | 10 | 248 | 6.5 ± 0.5 |
| 410P097 | HB | 307 | 27 | 256 | 9.2 ± 0.6 |
| 436P095 | HB | 290 | 25 | 241 | 9.4 ± 0.8 |
| 441P059 | HB | 282 | 17 | 241 | 7.3 ± 0.5 |
| 516P190 | HB | 329 | 21 | 284 | 8.1 ± 0.4 |
| 619P080 | HB | 314 | 20 | 270 | 8.6 ± 0.7 |
| 623P072 | HB | 313 | 14 | 275 | 8.6 ± 0.7 |

*Thickness error measured by the micrometer Mitutoyo (MDC-25SB) was 1 μm .

Table 3: Thicknesses of the Argisols sample port and samples prepared for OPC.

Comparing the thermal diffusivity determined by OPC with that obtained by TWI shows that both techniques agree in terms of the order of magnitude for the soils characterized. TWI is more sensitive to measurements in porous media compared to OPC because it minimizes the effects of air flow and pressure fluctuations within the pores. Another advantage of the TWI technique is that its theoretical model for opaque samples is described by non-transient parameters and can be applied over a wide frequency range for thermal wave oscillation. In contrast, the OPC model for thermal diffusion works only within a narrow frequency range and is subject to photobaric and

thermoelastic effects. However, using powdered samples for OPC was an effective method for capturing the photoacoustic signal from the soils [22].

3.3 Specific heat and density

Specific heat was obtained using the non-adiabatic thermal relaxation calorimetry technique, which involves heating and cooling cycles of the samples over time. Figure 10 shows the results of the specific heat measured for the various studied soils. Soils are heterogeneous materials composed of minerals, air, water, and organic matter. According to the de Vries model [35], the specific heat of soils is determined by the sum of the products of each constituent's specific heat and its corresponding volume fraction. Specific heat of water is $4180 \text{ J.Kg}^{-1}.\text{K}^{-1}$, of air is $1000 \text{ J.Kg}^{-1}.\text{K}^{-1}$ and that the specific heat known in the literature for hematite, goethite, sandy and silica are $c_{p\text{Fe}_2\text{O}_3} = 670 \text{ J.Kg}^{-1}.\text{K}^{-1}$, $c_{p\text{goethite}} = 800 \text{ J.Kg}^{-1}.\text{K}^{-1}$, $c_{p\text{sandy}} = 820 \text{ J.Kg}^{-1}.\text{K}^{-1}$, $c_{p\text{SiO}_2} = 730 \text{ J.Kg}^{-1}.\text{K}^{-1}$, respectively, and also organic matter [36, 37]. We can observe in the graph that organic matter changes the sensitivity of heat. In the subsurface horizons, more impoverished in organic compounds, the specific heat determined was established in the range from 936 to $1145 \text{ J.Kg}^{-1}.\text{K}^{-1}$, at the same time as for samples of horizon layer HA, specific heat was between 820 and $927 \text{ J.Kg}^{-1}.\text{K}^{-1}$.

Experimental measurements of specific heat for hematite and sandy soils showed good agreement with the literature. In horizon HA, the specific heat of the Nitosol and Latosol was slightly higher than that of the Argisol and Nitosol samples. In horizon HB, this increase was in the range of 10 to 20%. Soils are composed of approximately 50% of minerals and organic matter, and 50% of water and air. The higher moisture content in clayey soils, which have a more hydrophilic nature, may have contributed to their increased specific heat. Experimental errors were calculated using the standard deviation in the graph fitting software.

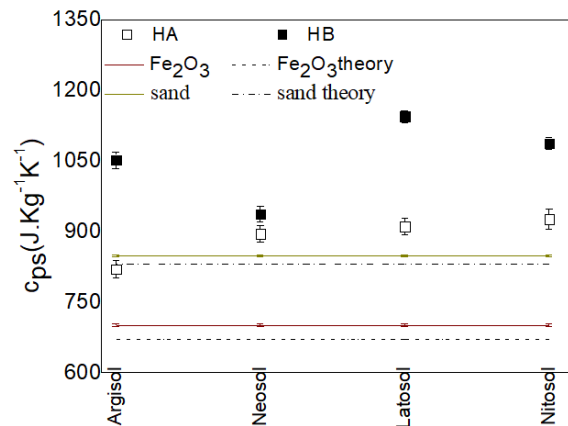


Figure 10: Specific heat of soils of the Parana, this figure also shows the predicted value of “sand” and for “iron oxide” (hematite).

Figures 11a and 11b show the bulk density and particle density of the soils. It is observed that both bulk density and particle density changed slowly with soil depth. Bulk density remained relatively stable at 1350 Kg.m^{-3} to clayey soils, and fluctuated between 1400 and 1900 Kg.m^{-3} to sandy samples, while the particle density was found from $(2212 \pm 190) \text{ Kg.m}^{-3}$ to (2494 ± 221)

Kg. m^{-3} and $(2527 \pm 216) \text{ Kg.m}^{-3}$ to $(2723 \pm 262) \text{ Kg.m}^{-3}$ in thin and thick textured soils, respectively.

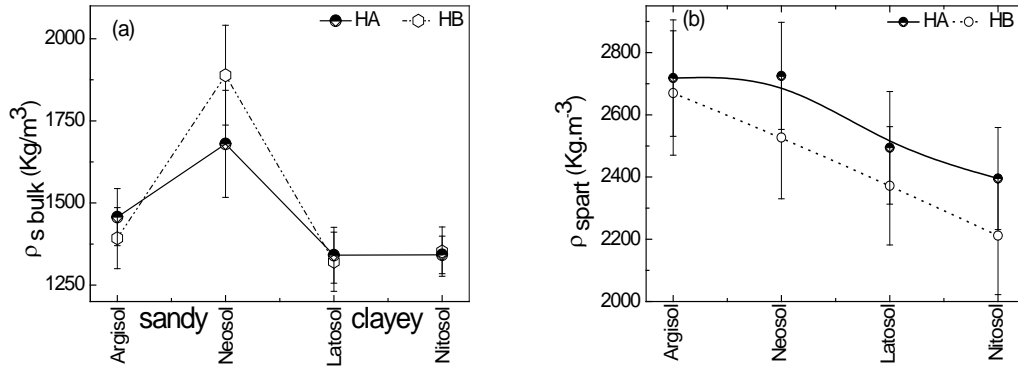


Figure 11a: Soils of Paraná bulk density and **Fig. 11b:** Soils Paraná particle density.

Density is associated with the compaction, structure, aggregation of organic matter, and porosity of the samples. Soil density is an important physical attribute for providing information on their conservation status control for infiltration and water retention, root development, gas exchange and susceptibility of this soil to erosion processes. Sand grains are approximately the same size as clay aggregates, but sand grains lack micropores. As a result, coarse-textured soils are denser than fine-textured soils. In clay soils, in addition to the porous space between aggregates, there are internal micropores. This structure provides more distributed porous space throughout the volume, leading to lower soil density. In sandy soils the concentration of organic matter is generally small, solid particles are little predisposed to cohesion and stickiness barring the clusters formation. Organic matter is hugely light weight floating between 200 and 400 Kg.m^{-3} and variable because of its complex composition, being altered by the decomposition time, wastes and residues. We found for natural sand (formed of 97% quartz, and wholly devoid of organic matter) an apparent density equal to 1997 Kg.m^{-3} , showing more dense than other soil samples with a sandy texture. While apparent density is based on ratio of mass and volume of dry soil, the bulk density only refers to the measurement of the specific mass of the solid fraction of the soils, while the density of particles is an unchanging property, independent of the dry and saturated soil conditions, the single attribute that significantly interferes in specific mass is the concentration of organic matter.

3.4 Final considerations about thermophysics properties soils

Results reported in literature indicated that the thermophysics properties determined in tests may be influenced by composition and amount organic matter present in soils. Soils from Parana are submitted to the tropical climate. Researches made on American, European, and Brazilian soils showed the properties found are influenced by a factors group, such as, soils in the natural state, dry, with fine and thick textures and extracted at different depths. As an example, Brazil's soils from the "caatinga" region (brushwood vegetation found in the Northeast of Brazil) were studied

and results from Silans [38] reported similar values for the thermal diffusivity of horizons HA (0 to 15cm). This region faces strong dry soils as well as soils interfaces with severe thermal gradient. In the opposite conditions, another reported result is found for the tundra ground in polar region. Such as very wet soils and low temperatures produces small thermal gradient in the soil's stratus. Bodzenta and coworkers [39] found values of thermal diffusivity from 7 to $34 \times 10^{-3} \text{ cm}^2/\text{s}$ for the horizon HA e HB (from 0 to 40 cm deep), with results representing typical properties for a mix soil with gravel, small stones, sand, clay and rocks, and they report the errors about 20% under the severe conditions from summer to winter in those location.

Reported values [27, 28] showed that thermal diffusivity (α_s) and the volumetric thermal capacity ($\rho_s c_{ps}$) from soil mixed with sand with humidity, and with the natural humidity, varying from 9 % to 38 %. Tests confirm that experimental results behaved satisfactorily with the prediction of the De Vries model [35]. Average thermal diffusivity and $\rho_s c_{ps}$ from soils with 9 % natural humidity were $9 \times 10^{-7} \text{ m}^2 \cdot \text{s}^{-1}$ and $1.66 \times 10^6 \text{ J} \cdot \text{m}^{-3} \cdot \text{K}^{-1}$. After the addition of water to each mm the liquid volume inside the soil slightly modified the behavior of thermal diffusivity and $\rho_s c_{ps}$, characterizing in a reduction of the diffusivity in the range from 9 to $8 \times 10^{-7} \text{ m}^2 \cdot \text{s}^{-1}$ and an increment of thermal capacity volumetric up to $6.25 \times 10^6 \text{ J} \cdot \text{m}^{-3} \cdot \text{K}^{-1}$. Employing the technique of the dual probe, Bristow and collaborators [28] characterized three American soils: Sandfly (100 % sandy), Clayton (98 % sandy) and Clarinda (clayey) dry, by determining the thermal diffusivity of the sandy soils $\alpha_s = 2.3 \times 10^{-7} \text{ m}^2 \cdot \text{s}^{-1}$ and $\alpha_s(\text{clarinda}) = 1.97 \times 10^{-7} \text{ m}^2 \cdot \text{s}^{-1}$. Soils volumetric thermal capacity was measured for Sandfly and Clayton as $1.2 \times 10^6 \text{ J} \cdot \text{m}^{-3} \cdot \text{K}^{-1}$ and $0.9 \times 10^6 \text{ J} \cdot \text{m}^{-3} \cdot \text{K}^{-1}$, respectively.

Nicolau and collaborators [40] characterized the thermal conductivity of the Yellow Latosol and the Neosol, using the Campbell model, obtaining, $k_s(\text{nitosol}) = 0.26 \text{ W} \cdot \text{m}^{-1} \cdot \text{K}^{-1}$, and $k_s(\text{yellow-argisol}) = 1.44 \text{ W} \cdot \text{m}^{-1} \cdot \text{K}^{-1}$. Temperature fluctuations and heat flux were monitored in Marajo Island between 2001 and 2005 [41], with the objective of comparing the thermal responses to daily heating. Thermal diffusivity and thermal conductivity were monitored during this period at depth levels, 0.00, 0.25 and 0.50 cm. Mean temperature obtained in the respective depths were, 303.3, 301 and 298 K, and the mean thermal diffusivity in the layers varied between 7.2 and $8 \times 10^{-7} \text{ m}^2 \cdot \text{s}^{-1}$ in the dry period, and from 7.9 to $28 \times 10^{-7} \text{ m}^2 \cdot \text{s}^{-1}$ during the rainy season [42] working with two soils types, classified as sandy and franco. Using the double probe method, the thermal conductivity found varied from 0.98 to $2.17 \text{ W} \cdot \text{m}^{-1} \cdot \text{K}^{-1}$ for the sandy soil and 0.51 to $0.78 \text{ W} \cdot \text{m}^{-1} \cdot \text{K}^{-1}$ for franco soil.

Characterization of multiple thermophysical data through the TWI contributes to the thermophysical mapping of the soils of Paraná and will allow in the future to draw a correlation with their respective bases analyzing the data via remote sensing, for example. Fluctuations of thermal parameters impact on all microbial activity of soils. Carbon dioxide concentration changes when the soil presents different temperatures [43]. Higher temperatures reduce the microbial activity of bacteria and fungi and increase the diversity of microbial community structure. Thermophysical properties also affect the rates of metal oxides.

Results indicated that the thermal characteristics of soil substrates lead to significant diurnal variations in soil temperatures. Profile analyses conducted by the authors using computerized X-ray tomography have demonstrated temporal variations in the soils. Using the OPC photoacoustic technique, Mota and collaborators [44] measured red clays collected from 1.2 to 2.5 m deep, in soil of the state of Rio de Janeiro (Brazil), it was pressed by 9 ton and thermally treated in the temperature range 300 to 1100 °C. Values of α_s were established between 1 to $4 \times 10^{-7} \text{ m}^2 \cdot \text{s}^{-1}$.

Momose and Kasubuchi [25] measured the thermal conductivity from three types of soil (Soil Ando, Soil Red Yellowy and Sand Toyoura) all soils of clayey texture, where the air pressure was reduced during experiment and temperature vary from 10 to 75 °C. Thermal conductivity increased markedly under reduced air pressure, beyond a critical soil water content of 101 kPa. The maximum thermal conductivity for each soil obtained was $2 \text{ W}\cdot\text{m}^{-1}\cdot\text{K}^{-1}$, such as manganese and iron in soils. Structural transitions are also affected by soil temperature [45]. In the models used in the simulations of heat transfer and temperature in the daily and seasonal cycles the thermophysical properties of thermal diffusivity, thermal conductivity and thermal effusivity are of important contribution. Tang and coworkers [46] reported that due to increased temperature during aerobic incubation at a concentration of organic matter were extremely reduced in soil samples. Jelinkhova and coworkers [47] canvassed the thermal regime and hydric in green roofs in very thin coverings of soils.

Photoacoustic, Photothermal, Heat Transient and Gravimetric techniques proved to be efficient for determination of thermophysical parameters. It is important to note that these are non-destructive, accessible techniques with good precision. The values obtained in this study showed good agreement with those found in the literature. The specific heat measured in the experiments corresponded well to the nature of the soils.

Apparent densities of agricultural land, which are essentially the Latosols and Nitosols, are in the range from 1200 to 1700 $\text{Kg}\cdot\text{m}^{-3}$, and the apparent densities of sandy soils correspond in the average to 1650 $\text{Kg}\cdot\text{m}^{-3}$ [36]. Apparent density of sandy found in the literature was 1750 $\text{Kg}\cdot\text{m}^{-3}$ and the bulk density is equal to 2500 $\text{Kg}\cdot\text{m}^{-3}$. Works in literature indicating that soils with apparent density superior than 1700 $\text{Kg}\cdot\text{m}^{-3}$ have serious difficulties for penetration from the roots, germination seeds, development and growth of plants because aeration and fast water vessel, because the diffusion of oxygen and other gases are changed, raising carbon gas concentration in the root area, and soils with apparent density higher than 2000 $\text{Kg}\cdot\text{m}^{-3}$ have the movement of water and transports of ions and nutrients fully inhibited.

The pursuit of sustainable practices has become increasingly important across various sectors, especially in construction and material industries. In this context, the thermal properties of materials such as thermal diffusivity, thermal conductivity, and thermal effusivity play a crucial role in energy efficiency and building sustainability [48]. This article explores the correlation between these thermal properties and sustainability, highlighting how understanding and applying these concepts can contribute in eco-friendly and economical practices.

Thermal conductivity is a material's ability to conduct heat. Materials with high thermal conductivity, such as metals, allow for rapid heat transfer, while materials with low thermal conductivity, such as foam, act as thermal insulators. In construction, choosing materials with low thermal conductivity can reduce the need for heating and cooling systems, thereby contributing to the energy efficiency of buildings and, consequently, to sustainability [49]. Thermal diffusivity measures how quickly heat spreads through a material. It is calculated from a material's thermal conductivity, density, and specific heat. Materials with high thermal diffusivity respond rapidly to temperature changes, which can be advantageous in extreme climates by helping to regulate indoor temperatures and reducing the need for artificial heating or cooling. Thermal effusivity is the capacity of a material to absorb and release heat [50]. Materials with high thermal effusivity can efficiently store and release heat, which can improve indoor thermal comfort. In terms of sustainability, materials with high thermal effusivity can help moderate temperature fluctuations, reducing the need for climate control systems and thereby decreasing energy consumption.

Understanding thermal properties is fundamental to sustainable agricultural practices like crop management, where choosing species suited to soil thermal conditions can optimize production and reduce the need for external inputs. Also in soil conservation, practices such as mulching can help regulate temperature and humidity, improving soil health and biodiversity and regularization of the environment temperature change. In the last case, rising global temperatures can affect the thermal properties of soils, impacting agriculture and environmental conservation.

The study of soils with appropriate thermal properties is very important for constructing sustainable buildings. Thermal conductivity, diffusivity, and effusivity play key roles in energy efficiency and thermal comfort, directly impacting building sustainability. Investing in materials that optimize these properties can lead to significant reductions in energy consumption, helping to build a more sustainable future.

4 Conclusion

The search for alternative detection systems that offer agility and cost reduction for measurements, with sensitivity to characterize various types of soils, including those affected by burning with higher salt concentrations, soils in erosion stages, exposure to toxic waste and contaminants, and samples from different climates such as semi-arid, temperate, cold, and those subjected to ice accumulation from snow has consistently motivated researchers. These systems also aid in monitoring the treatment and revitalization of soils.

In this work the technique TWI using white light produced measurements of temperature fluctuation in soils of different textures and horizons, making possible the characterization of thermal diffusivity, thermal effusivity and thermal conductivity. The results presented by TWI showed that in sandy soils whose predominant mineral constitution is formed by SiO_2 and $\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4$, are more conductive and thermally diffusive than clayey soils, of which its predominant mineral constitution are oxides Fe_2O_3 and $\text{FeO}(\text{OH})$. The increase in the thermal transport properties of the soils did not follow the same proportion as the predominant minerals. This is due to the presence of other secondary and primary minerals that also compose the soil, as well as the incorporation of existing organic matter in the porous fraction. Results indicated that in the surface horizon layer, poor in organic matter, the thermal diffusivity and thermal conductivity were lower than that in deeper horizons (subsurface). This result is important because it indicates that the heat tends to be distributed faster in soil surface rather than in bulk extracts. Samples were submitted to the same temperature change on the soils, which was about 2K in tests. Gravimetric measurements were also made to estimate the bulk and particle densities and gravimetric humidity from Parana soils. Pipette method was used to auxiliary in the soil densities identification. Bulk density and particle density of clayey soils were about 1350Kg.m^{-3} , and from 2300 to 2500Kg.m^{-3} , respectively. Soils bulk density sandy and medium sandy textures varied from 1400 to 1900Kg.m^{-3} and the particle density had range from 2500 to 2700Kg.m^{-3} . Lower-density soils are more porous because of their big cohesion of clay grains. By the bulk density measurements and heat specific by Thermal Relaxation method, it was possible to calculate the product of the volumetric thermal capacity and link it with the values achieved through TWI. Finally, the OPC system was used to measure soils as close as possible “in natural powder form” and characterize the thermal diffusivity of the Argisols in the horizons HA and HB. Results showed that the Argisols thermal diffusivity was smaller than $10^{-7}\text{m}^2.\text{s}^{-1}$. Data obtained for the thermophysical characterization of soils are related to simulations in heat transfer and water transfer models. This analysis examines the contribution of temperature to various soils, the effects

of organic material on temperature, and thermal anomalies in auxiliary soils. Additionally, it includes the generation of images through remote sensing for studies in various fields of knowledge.

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